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Analytical TEM study of the oxidation of nickel based superalloys

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Abstract. — The first stages of oxidation of different nickel based superalloys in air environment at 650 °C are investigated, using Transmission Electron Microscopy in association with analytical techniques. The alloys differ in microstructure (polycrystal or single crystal, type of hardening precipitates), and chemical analysis (effect of alloying elements). All experiments show evidence of a two oxide layer formation at the surface of the alloy. Different microstructural and chemical changes are demonstrated in the bulk near the metal/oxide interface, leading to a recrystallised band, depleted of hardening precipitates and resulting in a local mechanical softening.

1. Introduction.

A number of nickel based superalloys are known for being subject to intergranular or interfacial embrittlement due to oxidation. Earlier studies on fatigue crack growth at high temperature clearly demonstrate the influence of oxidation on crack propagation mode. Moreover, the damaging kinetics of the interfaces by oxidation is to be regarded as a major factor when considering high crack growth rates [1-5].

Since the early stages of oxidation determine the damaging process, it is important to characterize the different changes occurring in the material. This preliminary work excludes the mechanical aspects of such phenomena.

At medium temperatures (650 °C), oxidation mechanisms are driven by volume diffusion of elements, as well as interfacial and intergranular diffusion. A fine characterization of diffusion phenomena and an accurate chemical identification of oxidation scale growing at the surface of samples require a high resolution analytical tool.

A TEM, equipped with analytical spectrometers (EDX, EELS), meets these requirements, permitting one to monitor microstructural and chemical changes at a nanometric scale.

Different nickel based superalloys, differing in microstructure (monocrystal, polycrystal, precipitates) and chemical analysis (effect of alloying elements) were investigated after high temperature exposure under air environment.

2. Materials and experimental.

Two alloys were considered: first, a monocrystalline alloy. Its γ austenitic matrix is reinforced by a γ' phase, matrix coherent, precipitated as cubes (500 nm), in a 70 % volume ratio. Its

chemical composition is the following:

Ni	Co	Cr	Mo	W	Al	Ti	Ta	C	Zr
Bal	6,6	7,94	2,10	5,66	5,16	1,21	8,17	0,002	0,010

The behaviour of the γ/γ' interfaces during oxidation and the microstructure of the oxide scale were examined on this alloy.

Secondly, a polycrystalline alloy, Inconel 718, remarkable for its high Nb and Fe contents, and structural hardening by γ' and γ'' intermetallic phases. Its chemical composition is shown below:

Ni	Cr	Mo	Fe	Nb	Ti	Al	C
53,5	18,6	3,1	18,5	5	0,9	0,4	0,04

TEM investigation was carried out on a Philips EM430 equipped with a TRACOR energy dispersive X-ray spectrometer. The different phases and oxides have been identified by using electron diffraction techniques and concentration linescans in STEM mode, with a few nanometers probe size.

3. Results.

3.1 MONOCRYSTALLINE ALLOY. — In order to emphasize the importance of interfaces, thin foils, electrolytically thinned have been oxidized in air, at 650 °C, for various exposure times (30 secondes to 10 minutes). TEM examinations revealed differences in the oxidation mechanisms involved in the matrix, in the intermetallic phases and at the interfaces. After 10 min exposure (Fig. 1), γ' precipitates are covered with fine NiO grains, showing a polycrystalline growth of oxide, with no preferential cristallographic orientation to the substrate. Larger NiO crystallites (100 nm) have grown on the γ matrix, uniformly oriented in epitaxy as:

$$\begin{array}{l} [001]_{\gamma} // [001]_{\text{NiO}} \\ (001)_{\gamma} // (001)_{\text{NiO}} \end{array}$$

The interfaces exhibit a nanocrystalline area, devoid of NiO crystals. EDX analysis of these areas reveal a significant increase of Cr, Ta, Ti and Al contents at the interfaces, to the detriment of the matrix and the precipitates, as a complex oxide.

TEM observations of a cross-section specimen after 15 min oxidation in air at 650 °C (Fig. 2) reveals the presence of two distinct oxide layers on the surface of material. A concentration linescan across these oxides (Fig. 3) and electron diffraction patterns allowed their identification. The outer scale consisted of (Ni, Co)O grains, uniformly oriented. The inner scale was a complex nanocrystalline oxide containing Cr, Ta, Ti and Al. We assume that this complex oxide is at the origin of the CrTaO_4 , TiO_2 , AlTaO_4 mixture, reported by some authors for longer oxidation exposure times [6]. No composition gradient is observed in the metal near the interface metal/oxide.

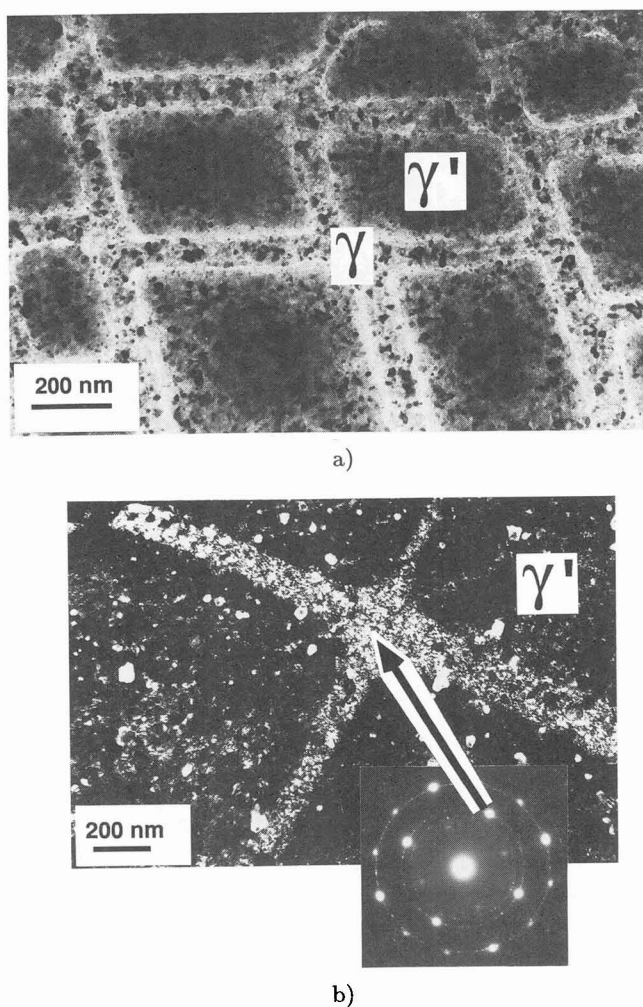


Fig. 1. — a) Bright field and b) dark field and diffraction pattern showing different oxidation mechanisms in the matrix, in the intermetallic phases and at the interfaces.

3.2 INCONEL 718. — TEM examination of cross-section samples oxidized in air 15 min at 650 °C (Fig. 4) and concentration linescan (Fig. 5) shows two distinct oxide layers. The outer scale is a NiO type oxide, showing no evidence of the presence of spinels, although containing Cr and Fe. The inner scale shows a variation in composition from Cr_2O_3 near the NiO interface, to $\text{Cr}(\text{Nb}, \text{Ti})\text{O}_4$ near the alloy.

The concentration linescan shows a depletion of Cr, Nb, Ti and Al in the matrix near the interface (50 % of initial concentrations). A dark field image from a substructure spot corresponding to $\gamma' + \gamma''$ indicates a disappearing of these precipitates in the same area (Fig. 6). The future evolution of this area has been checked with a 24 h oxidation under the same conditions, showing that this free precipitate area eventually results in a small grain size layer (Fig. 7).

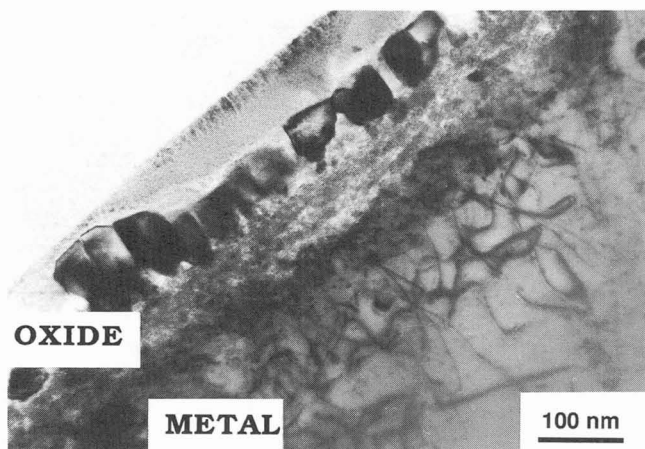


Fig. 2. — Monocrystalline alloy edge-on TEM observation after 15 min oxidation in air at 650 °C.

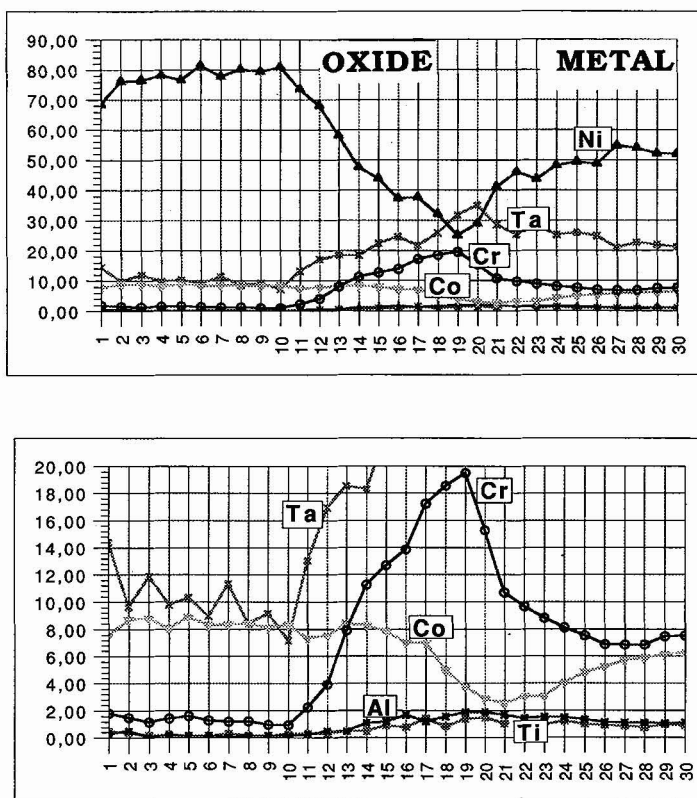


Fig. 3. — STEM elemental concentration linescan through the oxide/alloy interface (150 nm).

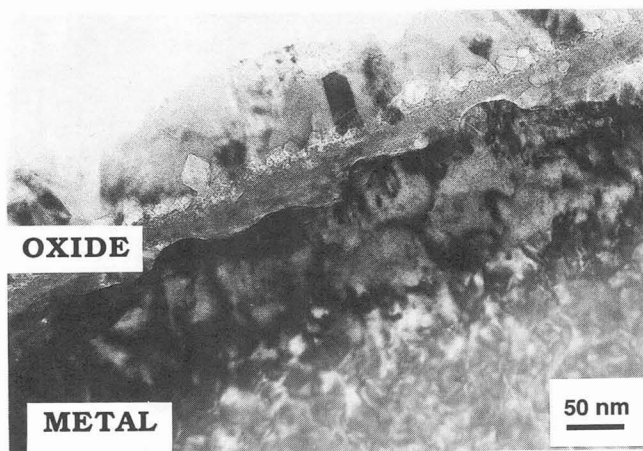


Fig. 4. — TEM edge-on examination of Inconel 718 oxidized in air 15 min at 650 °C.

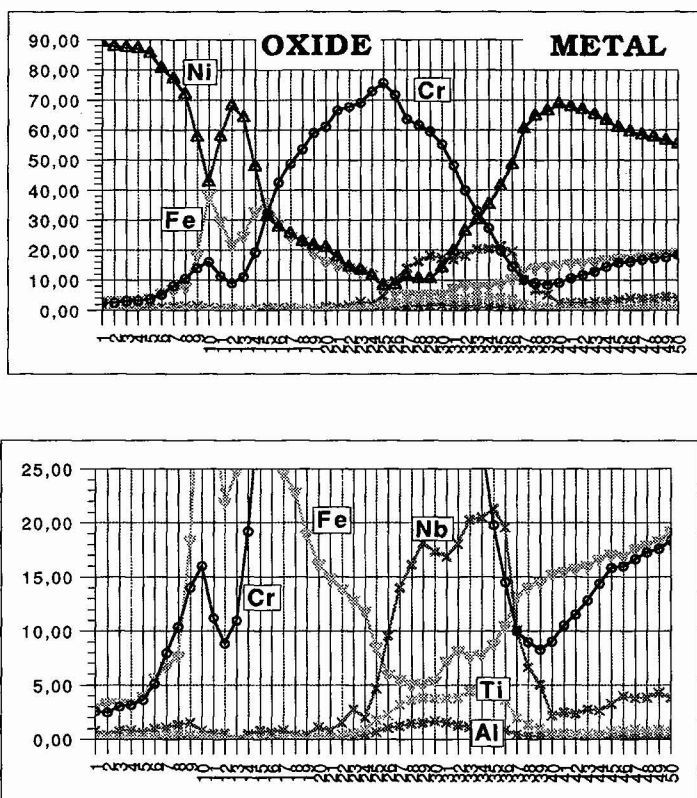


Fig. 5. — STEM elemental concentration linescan through the oxide/alloy interface (400 nm).

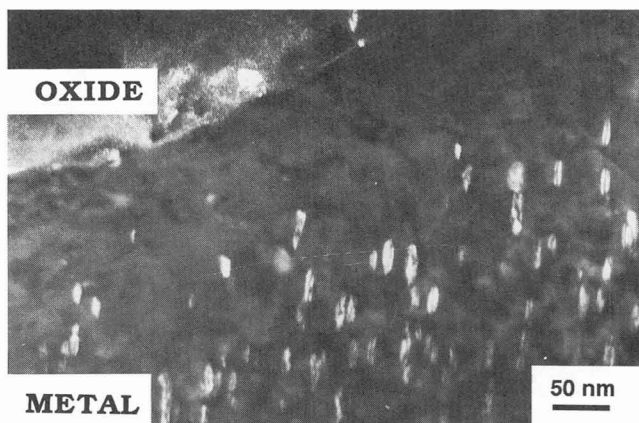


Fig. 6. — Dark field image revealing a γ'' depleted zone.

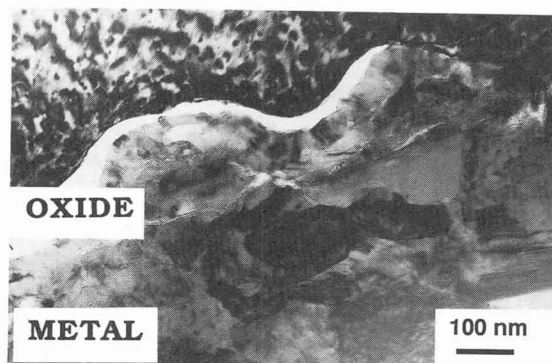


Fig. 7. — Sub grain structure beneath the interface.

4. Discussion and conclusion.

The monocrystalline alloy has been included in this study in order to establish a model of oxidation without any participation of grain boundaries. In this respect, further work has to be done on cross-section specimen with shorter exposure time. The first results showed evidence of a double layer oxide scale. The outer scale was NiO. Al, Ta, Ti and Cr diffuse to the interfaces to form a complex oxide which controls the further oxidation process.

Different microstructural and chemical changes of the Inconel 718 alloy have been demonstrated. In addition to the formation of two distinct oxide layers, a significant depletion of Cr content near the oxidized surface, together with a rapid disappearance of γ' + γ'' hardening precipitates is clearly noticeable. The high chromium content of this alloy lowers the solubility of Al and Nb in the nickel matrix so that it promotes Ni_3Al and Ni_3Nb precipitation. During oxidation, Cr diffuses to the surface to form Cr_2O_3 , and thus, the intermetallic phases are solutionized. This phenomenon is already well established for higher temperatures [6, 7].

This structural modification has an obvious influence on the mechanical behaviour of the

alloy. The existence of a superficial band, with lower Nb, Al and Ti contents leads to a local mechanical softening. The formation and growth of the oxide layer implies short range internal strains. A part of these strains is relaxed by plasticity increasing the dislocation content beneath the oxide. The combination of plasticity and softening causes recrystallization in this modified zone. The number of grain boundaries in this area is increased, enhancing Cr diffusion and thus provides a better capacity for passivation. Our examination of samples oxidized 15 min at 650 °C in air reveal Cr oxide. This implies that the passivation stage has been reached. This result has also been obtained indirectly using specific fatigue cycle with various hold time at minimum load [8].

The preliminary results of this study show the capability of TEM in association with microanalysis techniques for physico-chemical investigations requiring a nanometric scale resolution. The first stages of oxidation of superalloys result in very thin oxide scales and some chemical variations. Not all the experimental data have been explained, but at least all the phenomena taking place at the interfaces and in the bulk have been described, allowing for an interesting prospect of future results.

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